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1. It inhibits the release of a hormone like substance called prostaglandin 2. Regulates the elasticity of blood vessels and changes the function of blood platelets thus affecting blood clotting and easing inflammation.

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The Basic Chemistry of Acids and Bases. A lot of chemistry requires you to understand the difference between acids and bases. An acid is a substance that donates an H^+ ion to another chemical species called a base. A base is a substance that accepts (combines with) an H^+ ion.. If you need to know the

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concentration of the H^+ ion in solution, you can do so by representing the H^+ molarity, $[H^+]$.

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Questions from Chemistry 121 The mass of an electron is 9.11×10^{-31} kg if the de Broglie wavelength for an electron in a hydrogen atom is 3.31×10^{-10} m how fast is the electron moving relative to the speed of light the speed of light is 3.00×10^8 m/s

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1. each element is composed of tiny, indestructable particles called atoms
2. all atoms of a given element have the same mass and other properties that distinguish them from atoms of other elements
3. atoms combine in simple whole number ratios to form compounds
4. atoms of one element cannot change

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into atoms of another element

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Condensed electron configurations for the first 29 elements. 3. Be able to explain the difference between an atom and an ion? 4.

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does include examples and definitions from outcome 1.1 to 3.9. If you need assistance on further understanding a piece of the material, please email me at zcutl001@odu.edu. I hope this helps.

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Guide Chapters 1 3 Chapter 1 January 6
8 Matter and Energy What is chemistry
Chemistry is the study of the
composition structure and properties of
matter and of energy consumed or given
off when matter undergoes a change
Change a chemical reaction What were
some early uses for chemistry Fire
medicinal plants poisons for hunting

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Introduction to Chemistry 121
Classification of matter o What is matter

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Matter is anything that has mass and occupies space

o What are the two categories of matter Pure substances and mixtures

What is a pure substance A pure substance is composed of one type of matter examples table sugar pure water gold

A pure substance is either an element or compound

o What is an element An element is composed of one

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type of atom ...

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questions that assess the student's level of preparation in chemistry and basic math skills.

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