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Practice Test with
Answers. Question 1.
The standard EMF for

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the given cell reaction



E° is 1.10 V at 25 °C.

The EMF for the cell reaction, when 0.1 M Cu^{2+} and 0.1 M Zn^{2+} solutions are used, at 25 °C is. Related: Question Bank for NEET Physics.

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Answers 14. Questions
14-17 The spontaneous
reaction that occurs
when the cell in the
picture operates is as
follows: $2\text{Ag}^+ + \text{Cd (s)}$
 $\rightarrow 2 \text{Ag (s)} + \text{Cd}^{2+}$ (A)
Voltage increases. (B)
Voltage decreases but
remains $>$ zero.

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Questions 1. In an electrolytic cell the electrode at which the electrons enter the solution is called the _____; the chemical change that occurs at this electrode is called _____.

Sample Questions - Chapter 21

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Questions A 1. Which of these chemical

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reactions is an oxidation-reduction reaction? A. $\text{Fe} + \text{S} \hat{=} \text{FeS}$ C. $\text{AgNO}_3 + \text{NaCl} \hat{=} \text{AgCl} + \text{NaNO}_3$ B. $\text{CO}_2 + \text{H}_2\text{O} \hat{=} \text{H}_2\text{CO}_3$ D. $\text{H}_2\text{SO}_4 + 2\text{NaOH} \hat{=} \text{Na}_2\text{SO}_4 + 2\text{H}_2\text{O}$ C 2. What happens to the oxidizing agent in an oxidation-reduction reaction? A. It is oxidized as it gains electrons. C.

**APEF -
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Questions ...

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Questions (Type-II)

Note : In the following questions two or more than two options may be correct. The positive value of the standard electrode potential of $\text{Cu}^{2+} / \text{Cu}$ indicates that _____. (i) this redox couple is a stronger reducing agent than the $\text{H}^{+} / \text{H}_2$ couple. (ii) this redox couple is a stronger oxidising

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agent than H^+ / H_2 .

Questions

Class 12 Important Questions for

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Sample Questions -
Chapter 21

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Examples of Multiple
Choice Questions: 1. In
an electrolytic cell the
electrode at which the
electrons enter the
solution is called the
_____ ; the chemical
change that occurs at

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this electrode is called
 . (a) anode,
oxidation (b) anode,
reduction (c) cathode,
oxidation (d) cathode,
reduction

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Questions Solved**

31. During electrolysis of water the volume of oxygen liberated is 2.24 dm^3 . The volume of hydrogen liberated, under same conditions will be : 2.24 dm^3 .

1.12 dm^3 . 4.48 dm^3 . 0.56 dm^3 . Answer. 32.

In Down's method for the extraction of sodium, the melting point of the electrolyte is lowered by adding.

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Multiple choice questions. Try the following multiple choice questions to test your knowledge of this chapter. For each question there is one correct answer. The periodic table, physical constants and relative atomic masses needed for these problems are

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given on the inside covers of Chemistry, fourth edition by C.E. Housecroft and E.C. Constable. Once you have answered the questions, click on ...

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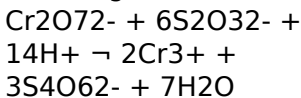
MULTIPLE CHOICE.

Choose the one
alternative that best
completes the
statement or answers
the question. 1) The
gain of electrons by an
element is called_____.

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- A) oxidation
 - B) sublimation
 - C) reduction
 - D) disproportionation
 - E) fractionation
- 2) _____

is reduced in the following reaction:



A.P. Chemistry
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A ...

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The aqueous solution
of which of the
following compounds is
the best conductor of
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