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contents: thermodynamics . chapter 01: thermodynamic properties and state of pure substances. chapter 02: work and heat. chapter 03: energy and the first law of thermodynamics. chapter 04: entropy and the second law of thermodynamics. chapter 05: irreversibility and availability

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Engineering Thermodynamics: Chapter-10 Examples. A Carnot vapor refrigeration cycle is used to maintain a cold region at 0 o F where the ambient temperature is 75 o F. Refrigerant R-134a enters the condenser as saturated vapor at 100 lbf/in² and

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leaves as saturated liquid at the same pressure. The evaporator pressure is 20 lbf/in². The mass flow rate of refrigerant is 12 lbm/s.

Engineering Thermodynamics: Problems and Solutions, Chapter-10

Engineering Thermodynamics: Chapter-5 Problems. A rigid tank of volume 10 m³ contains superheated steam at 1 MPa and 400 °C. Due to heat loss to the outside atmosphere, the tank gradually cools down to the atmospheric temperature of 25 °C. Determine (a) the heat transfer and (b) the entropy generated in the system's universe during this cooling process.

Engineering Thermodynamics: Problems and Solutions, Chapter-5

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1. INTRODUCTION Problem 1.3 Solution a) The mean velocity is $v_{ND} = 8k_B T / m$ where $k_B = 1.38 \times 10^{-23} \text{ J/K}$, $T = 273.16 \text{ K}$. The mass of the water molecule is $m = 18 \times 10^{-27} \text{ kg}$. The mean velocity is $v_{ND} = 6.022 \times 10^{23} \text{ mol}^{-1} \times 1.026 \text{ kg}$.

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SOLUTIONS THERMODYNAMICS PRACTICE PROBLEMS FOR NON-TECHNICAL MAJORS Thermodynamic Properties 1. If an object

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has a weight of 10 lbf on the moon, what would the same object weigh on Jupiter? Jupiter 22Moon c ft ft lbf-ft g =75 g =5.4 g =32 sec sec lbf-sec² c moon cmoon Jupiter Jupiter c mg Wg10×32 W = m = = 59.26 lb gg5.4 mg 59.26×75 W = 139 ...

Thermodynamic Properties

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