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General Chemistry Interchapters Online Fourth Edition Donald A. McQuarrie, Peter A. Rock and Ethan B. Gallogly

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8 GENERAL CHEMISTRY, FOURTH EDITION | McQuarrie, Rock, and Gallogly 4-5. We first use Equation 4.1  $v = c / \lambda = 2.9979 \times 10^8 \text{ m}\cdot\text{s}^{-1} / 656.3 \times 10^{-9} \text{ m} = 4.568 \times 10^{14} \text{ s}^{-1}$  The energy is given by Equation 4.3  $E = h\nu = (6.626 \times 10^{-34} \text{ J}\cdot\text{s})(4.568 \times 10^{14} \text{ s}^{-1}) = 3.027 \times 10^{-19} \text{ J} = 0.3027 \text{ eV}$  4-6. We use Equation 4.7  $\lambda = h / mv = 6.626 \times 10^{-34} \text{ J}\cdot\text{s} / (45.9 \text{ g}) / (1 \text{ kg} / 1000 \text{ g}) / (120 \text{ mi}\cdot\text{h}^{-1})$

**CHAPTER 1. Chemistry and the Scientific Method**

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