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alCulations and
equations 10, that can
be made from 109 ...

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CHAPTER 1 REVIEW
Matter and Change
MIXED REVIEW SHORT
ANSWER Answer the
following questions in
the space provided 1
Classify each of the
following as a

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campwallamazu. Terms
in this set (38) kinetic-
molecular theory.

Based on the idea that
particles of matter are
always in motion. ideal
gas. A hypothetical gas
that perfectly fits all
the assumptions of the
kinetic-molecular
theory.

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CHAPTER 10 REVIEW
States of Matter

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SECTION 3 SHORT

ANSWER Answer the

following questions in
the space provided. 1.

Match description on
the right to the correct
crystal type on the left.

b ionic crystal (a) has
mobile electrons in the
crystal c covalent

molecular crystal (b) is
hard, brittle, and
nonconducting a

metallic crystal (c)
typically has the lowest
melting point of the

four

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Green, Michael / Modern Chemistry Textbook

3 at 25°C is 10. g of solute per 100. g of H

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20. a. If 15 g of KClO_3 are stirred into 100 g of water at 25°C , how much of the KClO_3 will dissolve? Is the solution saturated, unsaturated, or supersaturated? 10 g of KClO_3 will dissolve, but 5 g will not, despite thorough stirring. The solution is saturated.

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Chapter Test Name
Class Date Chapter
Test B, continued Refer
to the following
statement to answer
questions 30-32: Dilute
 $\text{HCl}(\text{aq})$ and $\text{NaOH}(\text{aq})$
are mixed in
chemically equivalent
quantities.

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CHAPTER 7 REVIEW
Chemical Formulas and
Chemical Compounds

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MIXED REVIEW SHORT ANSWER Answer the following questions in the space provided.

1. Write formulas for the following compounds:

CuCO₃ a. copper(II) carbonate
Na₂SO₃ b. sodium sulfite
(NH₄)₃PO₄ c. ammonium phosphate
SnS₂ d. tin(IV) sulfide
HNO₂ e. nitrous acid

2.

7 Chemical Formulas and Chemical Compounds

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Change 2 1 2 3
Chemistry Is a Physical
Science 3 Matter and
Its Properties 6
Elements 16 ... two
chemicals are mixed.
Sometimes, scientists
do basic research
simply to satisfy their
curiosity about a
chemical and its
qualities. Sometimes

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structural detail makes
a Water & Solutions -
for Dirty Laundry:
Crash Course
Chemistry #7
Dihydrogen monoxide
(better know as water)
is the key to nearly

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everything It falls ...

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Answers

CHAPTER 1 REVIEW
Matter and Change
MIXED REVIEW SHORT
ANSWER Answer the
following questions in
the space provided. 1.
Classify each of the
following as a
homogeneous or
heterogeneous

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substance. a. sugar d.
plastic wrap b. iron
filings e. cement
sidewalk c. granola bar
2. For each type of
investigation, select
the most appropriate
branch of chemistry
from the following

mc06se cFMsr i-vi

An easier way to
understand reactions is
to once again consider
cooking, but this time,
think about making a
pizza Modern

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chemistry chapter 13
Review Answers.

The foundation of pizza is, of course, the crust, and pizza dough is the raw . . .

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equations 10, that can
be made from $109 \times$

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104 kilograms of
phosphorus in the
second of these ...

CHAPTER 1 REVIEW

Matter and Change

MIXED REVIEW SHORT

ANSWER Answer the

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Chapter Test Chapter:
Acid-Base Titration and
pH PART I In the space
provided, write the
letter of the term or
phrase that best
completes each
statement or best
answers each question.

_____ 1. The pH scale
generally ranges from
a. 0 to 1. b. 1 to 1. c. 0
to 7. d. 0 to 14. _____ 2.

**Assessment Chapter
Test B - marshallh.**

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