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The first step in the chain-reaction polymerization process, initiation, occurs when the free-radical catalyst reacts with a double bonded carbon monomer, beginning the polymer chain. The double carbon bond breaks apart, the monomer bonds to

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the free radical, and the free electron is transferred to the outside carbon atom in this reaction.

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Polymerization is autoaccelerated because of large changes in viscosity during polymerization. Living polymerization by atom transfer, nitroxides, and addition-fragmentation transfer allow the synthesis of block, graft, star, and other polymer architectures. Increased pressure alters polymerization rate and polymer molecular weight.



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Abstract Addition polymerization of alkenes is an exothermic process due to the relief of the bond strain from C-C double bonds ( $\pi$  bonds) in the monomer molecules. Control of exothermic reactions is crucial for preventing runaway reactions/explosions in the industrial setting.

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